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Entropy of Mixing is Just the Entropy of
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A full discussion of the molecular basis
for thermodynamics is Statistical
Mechanics. Consider two isomers, i.e.

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two compounds that have the same molecular formula but a different molecular structure. Let us assume for the moment that the two isomers have the same electronic energy in the lowest or ground state.

AN INTRODUCTION TO THERMODYNAMICS

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proportional to the average thermal (ie,
random) translational kinetic energy of
molecules, with the constant of
proportionality such that: $\frac{1}{2} m \overline{v^2} = \frac{3}{2} k_B T$... [Book]

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Statistical mechanics, also called

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statistical thermodynamics, emerged with the development of atomic and molecular theories in the late 19th century and early 20th century, and supplemented classical thermodynamics with an interpretation of the microscopic interactions between individual particles or quantum-mechanical states.

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thermodynamics of ionic mixtures at the

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undergraduate and graduate level.

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Our goal in this chapter is to extend the concepts of thermochemistry to an exploration of thermodynamics (from the Greek thermo and dynamic, meaning “heat” and “power,”

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respectively), the study of the interrelationships among heat, work, and the energy content of a system at equilibrium. Thermodynamics tells chemists whether a particular reaction is energetically possible in the ...

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